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## REPORT DOCUMENTATION PAGE

Form Approved OMB No. 0704-0188

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PLEASE DO NOT RETURN YOUR FO	ORM TO THE ABOVE ORGANIZATION	N.		
1. REPORT DATE (DD-MM-YYYY) 2. REPORT TYPE		3. DATES COVERED (From - To)		
06-09-2008	Final Report		From 04-01-2007 To 03-31-2008	
4. TITLE AND SUBTITLE		1	NTRACT NUMBER	
Reaction mechanisms and Velocity in Dense, Layered, nanoenergetic materials			FA9550-07-1-0349	
		5b. GR	5b. GRANT NUMBER	
			5c. PROGRAM ELEMENT NUMBER	
		5c PPC		
		J. 1 K		
6. AUTHOR(S)			5d. PROJECT NUMBER	
Coffey, Kevin R.				
		5e. TAS	5e. TASK NUMBER	
		Sf WOS	5f. WORK UNIT NUMBER	
31.		31. 110.	WORK ONLY NOMBER	
7. PERFORMING ORGANIZATION N.	AME(S) AND ADDRESS(ES)		8. PERFORMING ORGANIZATION REPORT NUMBER	
University of Central Florida			REPORT NUMBER	
4000 Central Florida Blvd.				
Orlando, FL 32816	T T			
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SPONSORING/MONITORING AGENCY NAME(S) AND ADDRESS(ES)  Air Force Office of Scientific Research				
/ In Force Office of Scientific Resource			AFOSR	
			11. SPONSOR/MONITOR'S REPORT	
			AFRL-OSR-VA-TR-2013-0953	
			AFRE-OSK ***	
12. DISTRIBUTION/AVAILABILITY STATEMENT				
13. SUPPLEMENTARY NOTES				
14. ABSTRACT				
This project served to extend the development of nanoscale thermite-based energetic materials and consisted of the four following tasks which have				
been completed.				
oven completed.				
I) Develop an optical time-of-flight reaction velocity measurement technique suitable for use with free-standing nanolayered thin films.				
2) Continue examination of the Al/CuO materials system to assess an empirical upper limit for the reaction velocity as a function of thin film				
processing and layer structure.				
3) Provide samples to collaborators, for further characterization of the reaction process and mechanisms.				
4) Continue the development of sample handling procedures for these materials.				
15. SUBJECT TERMS				
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16. SECURITY CLASSIFICATION OF:  17. LIMITATION OF  18. NUMBER  19a. NAME OF RESPONSIBLE PERSON  OF  OF				
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Grant/Contract Title: Reaction mechanisms and Velocity in Dense, Layered, nanoenergetic materials

Grant/Contract Number: FA9550-07-1-0349

This project consisted of the four following tasks which have been completed.

Develop an optical time-of-flight reaction velocity measurement technique suitable for use with free-standing nanolayered thin films. This task has been completed, and measurement data are shown in figure 1. This sample exhibits an average propagation velocity of 161 m/sec, confirming previous high speed photography measurements of freestanding films of  $150 \pm 30 \text{ m/sec}$  wherein the accuracy was limited by the photographic frame rate.

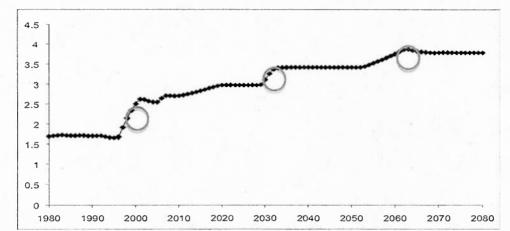


Figure 1: Reaction of a free-standing film sample with the [Al 26nm/CuO 54nm]<sub>40</sub> structure Peak 1-to-2 is 156.25 m/s and peak 2-to-3 is 166.67 m/s

- 2) Continue examination of the Al/CuO materials system to assess an empirical upper limit for the reaction velocity as a function of thin film processing and layer structure. Both poster and powerpoint presentations of the results of our continued examination of these materials were presented during at the AFOSR Nanoinitiative Meeting, May 5-7, 2008 in Fairborn, Ohio. These presentations are attached.
- Provide samples to collaborators, for further characterization of the reaction process and mechanisms. These samples were prepared for shipment and the MSDS documentation to enable receipt was also generated and provided. Evaluation of these materials is pending the collaborator's permission to ship, which has been delayed as their experiment has been delayed.
- Continue the development of sample handling procedures for these materials.

  For the highest performance samples with the [Al 26nm/CuO 54nm]<sub>40</sub> structure, the initial yield was slightly less than 30% with the losses (due to electrostatic discharge events) evenly distributed between losses during deposition and post deposition handling. Post deposition losses were brought to a minimum through keeping the material at ground potential. Deposition losses were reduced by increasing the conductivity of the sample holder by pre-coating with copper and the mechanical grounding of the systems transfer mechanisms. These changes improved the yield for these structures to 75%, which will enable further development of samples having a higher reaction velocity.

## Final Progress Survey Profile Report

Date Published: 09/08/2008

## Page One

1. Principal InvestigatorName:

Kevin R. Coffey

2. Grant/Contract Title:

Reaction mechanisms and Velocity in Dense, Layered, nanoenergetic materials

3. Grant/Contract Number:

FA9550-07-1-0349

4. Reporting Period Start (MM/DD/YYYY):

04/01/2007

5. End (MM/DD/YYYY):

03/31/2008

6. Program Manager:

Michael Berman

7. Annual Accomplishments (200 words maximum):

This project served to extend the development of nanoscale thermite-based energetic materials and consisted of the four following tasks which have been completed.

- 1) Develop an optical time-of-flight reaction velocity measurement technique suitable for use with free-standing nanolayered thin films.
- 2) Continue examination of the Al/CuO materials system to assess an empirical upper limit for the reaction velocity as a function of thin film processing and layer structure.
- 3) Provide samples to collaborators, for further characterization of the reaction process and mechanisms.
- 4) Continue the development of sample handling procedures for these materials.
- 8. Archival Publications (published) during reporting period:

None

9. Changes in research objectives (if any):

None

10. Change in AFOSR program manager, if any:

None

11. Extensions granted or milestones slipped, if any:

No-cost extension was granted to extend the project end date from 11/31/207 to 03/31/2008

12. Attach Final Report (max. 2MB)(If the report is larger than 2MB, please email file to program manager.)

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13. Please attach saved SF298 Form here: (Please be sure to have already saved the SF298 Form, that you plan to attach to this survey, to your desktopso that it may be uploaded within this field.)

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